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Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Vertical Electronic Spectra of the Isovalent Molecules H₂CNH, H₂SiNH, H₂CPH and H₂SiPH on the Basis of MRD-CI Calculations

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To cite this Article Bruna, Pablo J., Krumbach, Volker and Peyerimhoff, Sigrid D.(1987) 'Vertical Electronic Spectra of the Isovalent Molecules H₂CNH, H₂SiNH, H₂CPH and H₂SiPH on the Basis of MRD-CI Calculations', Phosphorus, Sulfur, and Silicon and the Related Elements, 30: 3, 698

To link to this Article: DOI: 10.1080/03086648708079192 URL: http://dx.doi.org/10.1080/03086648708079192

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Vertical Electronic Spectra of the Isovalent Molecules H₂CNH, H₂SiNH, H₂CPH and H₂SiPH on the Basis of MRD-CI Calculations

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Large-scale multi-reference single and double-excitation configuration interaction (MRD-CI) calculations are employed for the study of the isovalent compounds H₂CNH, H₂SiNH, H₂CPH and H₂SiPH in their ground state equilibrium geometry. The dipole moments and charge distributions are given. The vertical excitation energies to the intra-valence states $^{3,1}(\text{n,\pi}^{\,*})$ and $^{3,1}(\text{\pi}^{\,},\text{\pi}^{\,*})$ and to the first members of the Rydberg series originating from n and π MO's respectively are predicted; the first two ionization potentials and the Rydberg term values are also calculated. In H2CNH, mixing of Rydberg and valenceshell states with CN stretching is analyzed. The trends in relative stability of electronic and ionized states can be directly related to increased orbital stability of n relative to π as soon as a first-row constituent is replaced by a second-row atom. The calculations explain the diffuse character of the uv spectrum of imines; they treat the molecules ${
m H}_2{
m SiNH}$ and ${
m H}_2{
m SiPH}$ for the first time and present a large number of data for all four molecules which can serve as a basis for future experimental investigations on these and related compounds.